

The α/β Interfaces of $\alpha_1\beta_1$, $\alpha_3\beta_3$, and F_1 : Domain Motions and Elastic Energy Stored during γ Rotation¹

Yasuo Kagawa,^{2,3} Toshiro Hamamoto,² and Hitoshi Endo²

ATP synthase (F_0F_1) consists of F_1 (ATP-driven motor) and F_0 (H^+ -driven motor). F_1 is a complex of $\alpha_3\beta_3\gamma\delta\epsilon$ subunits, and γ is the rotating cam in $\alpha_3\beta_3$. Thermophilic F_1 (TF_1) is exceptional in that it can be crystallized as a β monomer and an $\alpha_3\beta_3$ oligomer, and it is sufficiently stable to allow $\alpha\beta$ refolding and reassembly of hybrid complexes containing 1, 2, and 3 modified α or β . The nucleotide-dependent open–close conversion of conformation is an inherent property of an isolated β and energy and signals are transferred through α/β interfaces. The catalytic and noncatalytic interfaces of both mitochondrial F_1 (MF_1) and TF_1 were analyzed by an atom search within the limits of 0.40 nm across the $\alpha\beta$ interfaces. Seven (plus thermophilic loop in TF_1) contact areas are located at both the catalytic and noncatalytic interfaces on the open β form. The number of contact areas on closed β increased to 11 and 9, respectively, in the catalytic and noncatalytic interfaces. The interfaces in the barrel domain are immobile. The torsional elastic strain applied through the mobile areas is concentrated in hinge residues and the P-loop in β . The notion of elastic energy in F_0F_1 has been revised. X-ray crystallography of F_1 is a static snap shot of one state and the elastic hypotheses are still inconsistent with the structure, dynamics, and kinetics of F_0F_1 . The domain motion and elastic energy in F_0F_1 will be elucidated by time-resolved crystallography.

KEY WORDS: ATP synthase; F_1 ; elastic energy; domain motion; hinge; F_0F_1 ; α/β interface; rotation thermophilic; X-ray crystallography.

¹ Key to abbreviations: F_0F_1 , ATP synthase; F_1 , catalytic portion of F_0F_1 ; F_0 , ion-conducting portion of F_0F_1 ; MF_1 , TF_1 , and EF_1 , F_1 from mitochondria, thermophilic bacillus PS3, and *E. coli*, respectively; $\alpha_3\beta_3\gamma\delta\epsilon$, subunits of F_1 ; DCCD, dicyclohexylcarbodiimide; DELSEED, -Asp-Glu-Leu-Ser-Glu-Glu-Asp- sequence in the N-terminal region of β (in TF_1 , it is DELSDED); loop AB, loop connecting the flanking secondary structures A and B; P-loop, -Gly-X-X-X-Gly-Lys-Thr-sequence of both the α and β subunit of F_1 ; 2,8-diN₃-ATP, 2,8-diazido-ATP; SL-ATP, spin-label ATP; AMP-PNP, 5'-adenylylimido diphosphate; Bz-ADP, 3'-O-(4-benzoyl) benzoyl-ATDP; OSCP, oligomycin-sensitivity conferring protein; OOO, α_3 (open β)₃ form; OOC, α_3 (open β)₂ (closed β) form; OCC, α_3 (open β) (closed β)₂ form; CCC, α_3 (closed β)₃ form. An example of residue nomenclature: T β E190, Glu190 (E190) residue of the $TF_1\beta$ subunit. I–XI, area of contact in the α/β interfaces.

² Department of Biochemistry, Jichi Medical School, Minamikawachi, Tochigi, 329-0498, Graduate School, Women's University of Nutrition, Sakado, Saitama 350-0288, Japan.

³ To whom all correspondence should be addressed: e-mail: ykagawa@jichi.ac.jp or kagawa@sak.eiyo.ac.jp

In review, published in this journal (Kagawa and Hamamoto, 1996), we supported the rotational hypothesis of ATP synthase (F_0F_1) (Boyer, 1997 and references therein) and predicted that the torque of the rotation is 42 pN·nm via elastic energy transiently stored in F_0F_1 (Kagawa, 1999). In that issue, the Editor recommended that methodologies be developed that will allow subunit movements to be monitored as a function of time and captured directly on film (Pedersen, 1996). In fact, the intramolecular rotation of γ in the fixed $\alpha_3\beta_3\gamma$ of thermophilic F_1 (TF_1) has been visualized using single-molecule imaging (Noji *et al.*, 1997; Yasuda *et al.*, 1998; Kinoshita, 1999), and the rotation theory was established. The unsolved question is the analog–digital conversion of energy in F_0F_1 . The electrochemical energy of the H^+ current (Mitchell, 1979) through F_0 in liposomes (Kagawa and Racker, 1971) and a planar bilayer (Hirata *et al.*, 1986), and the mechanical energy of the rotation are analog quantities. The number of ATP molecules released from

the catalytic site is a digital quantity. Despite the analog-digital conversion, F_0F_1 shows a fixed stoichiometry and nearly 100% energy efficiency (Yasuda *et al.*, 1998). To convert the $\pi/5$ step rotation of F_0 into the $2\pi/3$ step rotation of F_1 , we predicted elastic energy (Kagawa and Hamamoto, 1996). Since rotation occurs during the time lag that follows the hydrolysis of ATP, there must be transient elastic energy stored in F_0F_1 (Kagawa, 1999).

Recently, the elastic energy of F_0F_1 has been calculated on the hypothetical "four-step stress accumulation" (Cherepanov *et al.*, 1999) or "binding zipper" (Oster and Wang, 2000). However, the following experimental data are inconsistent with their calculation.

1. It is based on the graphical rotation (<http://nature.berkeley.edu/~hongwang/ATP-synthase>) of X-ray crystallography of F_1 (Abrahams *et al.*, 1994). However, it is a static snap shot of one state during rotation and ATPase is inactive in that crystallizing condition (high Mg^{2+} , ADP, and AMP-PNP).

2. The reported number of c subunits in F_0 is 12 (Jones *et al.*, 1998), whereas the X-ray crystallography indicates that this number is 10 (Stock *et al.*, 1999). Thus, F_0 is rotated by the $\pi/5$ step, not the $2\pi/3$ step rotation of F_1 that is used for their calculation. The single-molecule analysis of rotating F_0F_1 (Sambongi *et al.*, 1999) was refuted because of defective F_0 (Tsunoda *et al.*, 2000).

3. In contrast to the X-ray crystallography (Abrahams *et al.*, 1994), all sites are fully occupied by Mg-nucleotides during the steady state (Weber and Senior, 2000). The fully occupied F_1 was also suggested by X-ray crystallography (Bianchet *et al.*, 1998).

4. The calculation is based on the unisite catalysis at the ATP concentration of 1 mM (Fig. 4 of Ref. Oster and Wang, 2000). However, the F_1 rotation is observed at the bi- and trisite catalysis (Yasuda *et al.*, 1998).

5. TF_1 is harder than mesophilic F_1 s. The strain-stress relationship should be different, yet the synthesis/hydrolysis of ATP is identical (Kagawa and Hamamoto, 1997).

6. Although γ is a putative torsional spring according to Cherepanov *et al.* (1999), it is not limited to γ but is distributed among residues including hinge-controlling interfaces of subunits.

7. Tritium exchange (Noji *et al.*, 1996) and IR (Ohta *et al.*, 1980) failed to detect "the zippering of H bonds" (Oster and Wang, 2000) in both TF_1 and γ during ATP hydrolysis. The long-reaching electrostatic force is more important than the H bond (0.3 nm, directional) and van der Waals forces (short range) in elastic domain displacement.

Such elastic strain must be transferred through the subunit interfaces into a heterogeneous solid composed of

the α -helices, β sheets, and loops. The molecular structure of the ATP-binding site located at the α/β interface has been visualized by X-ray crystallography (Abrahams *et al.*, 1994; Shirakihara *et al.*, 1997; Bianchet *et al.*, 1998). F_1 s from thermophilic bacteria (T) (Kagawa *et al.*, 1986), mitochondria (M) (Kagawa *et al.*, 1997), and *E. coli* (E) (Futai *et al.*, 1989), are called TF_1 , MF_1 , and EF_1 , respectively, and have various advantages in the structure study.

This review analyzes the molecular dynamics of the α/β -subunit interfaces and hinges based upon data obtained by Yoshida's group and ours. Contact residue pairs within 0.40 nm across the α/β interfaces in $\alpha_3\beta_3$ complex (Shirakihara *et al.*, 1997) and MF_1 (Abrahams *et al.*, 1994) were computed using the CCP4 Suite: Program Contact. Detailed structures of $T\beta$ can be viewed via the internet at NCBI MEDLINE PubMed MMDB Id:7378 PDB Id:1SKY. Detailed mutational studies on EF_1 have been performed (Weber and Senior, 2000; Futai *et al.*, 1989) and the proposed imaginary roles of the residues will be examined in the light of physical data on TF_1 . Open and closed conformations are known for β and the theoretical treatment of the domain motions of 24 proteins with two conformers has been reported (Hayward, 1999).

STATOR AND ROTOR IN F_0F_1

Components of F_0 and F_1 (Fig. 1)

We prepared and reconstituted MF_0F_1 from MF_0 and MF_1 (Kagawa and Racker, 1966a) and then assembled TF_1 from refolded subunits (Yoshida *et al.*, 1975). The subunit compositions of F_1 and prokaryotic F_0 are $\alpha_3\beta_3\gamma\delta\epsilon$ and ab_2c_{10} , respectively. The $\gamma\epsilon$ subunits of F_1 rotate in the cavity of $\alpha_3\beta_3$ fixed by δb_2 to F_0 - a subunit. F_1 is about 370 kDa, and the sequence of α (55 kDa), β (52 kDa), and γ (30–32 kDa) subunits are homologous among F_1 s of all species. The isolated α and β both have AT(D)P binding sites. In MF_0F_1 , OSCP, and δ corresponds to δ and ϵ , respectively, of prokaryotic F_1 . MF_0 contains, in addition to the common a , b , and c , specific subunits d , e , f , g , F_6 , A6L, and OSCP (Collinson *et al.*, 1996).

Single-Stalk F_0F_1 and Rotation

F_0F_1 was identified as a sphere (diameter 12×10 nm, $\alpha_3\beta_3$) connected by two stalks (central and peripheral) to a basal piece (Wilkins and Capaldi, 1998). The central stalk contains $\gamma\epsilon$, and the peripheral stalk, δ - F_0 - b (Fig. 1, lower right). The $\gamma\epsilon$ is the eccentric shaft

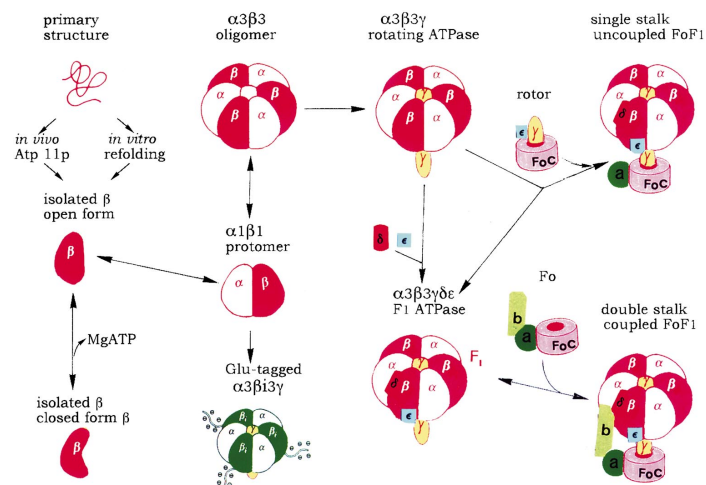


Fig. 1. Refolding and assembly of subunits of thermophilic F₁. The nomenclature of subunits is described in the text. F₁ subunits are α , β , γ , δ , and ϵ . F_o subunits are a , b , and c . Details have been reported (Kagawa, 1999).

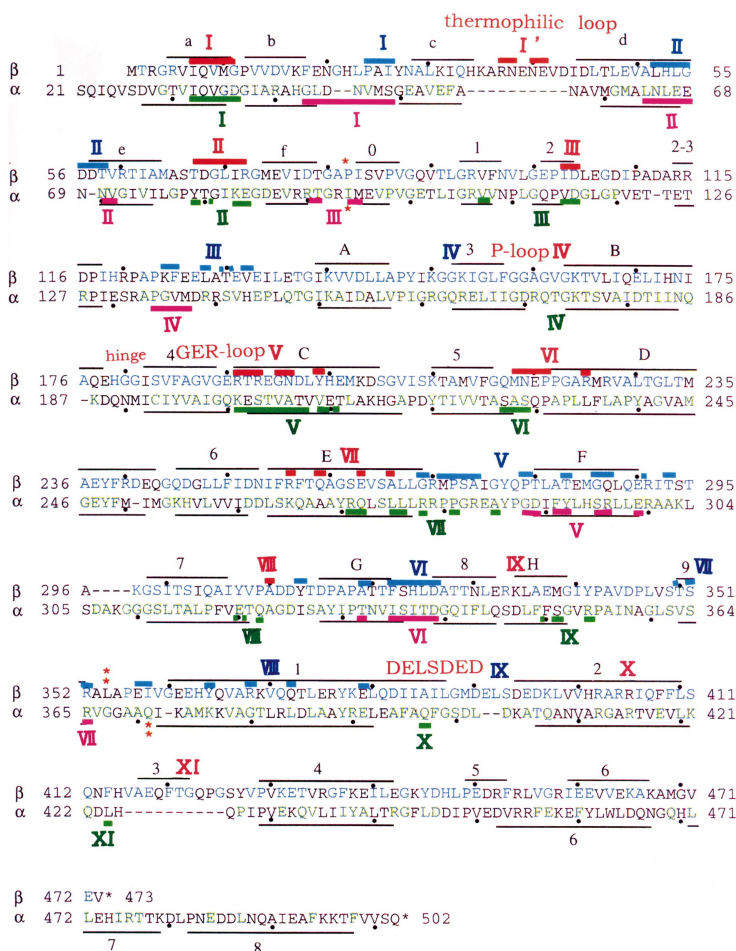


Fig. 2. Catalytic and noncatalytic $\alpha\beta$ interfaces located on aligned amino acid sequences and secondary structure elements of α and β subunits in TF₁. Solid black lines indicate folds and these are classified into α -helices (A–H, 1–8) and β -sheets (a–f, 0–8). Folds of both subunits are similar and labels are thus provided only for the β subunit, except for the three C-terminal α -helices in the α subunit. Dots indicate every tenth residue. I–XI: areas of $\alpha\beta$ contact. Red, catalytic contact areas of β ; Pink, catalytic contact areas of α ; blue, noncatalytic contact areas of β ; green, noncatalytic contact areas of α . Colored bars indicate contact residues in T β E. Sequences are divided by red asterisks to indicate the three domains. Original sequence data (Kagawa *et al.*, 1986) are aligned as described (Kagawa, 1999).

and δ -F₀-b is the part of stator that fixes $\alpha_3\beta_3$ to F₀-a. In this double stalk F₀F₁ torque of the $\gamma\varepsilon$ -F₀-c rotor induces a conformational change of $\alpha\beta$ (Fig. 1).

The single molecule analysis of rotating F₀F₁ (Sambongi *et al.*, 1999) is controversial because of defective F₀ (Tsunoda *et al.*, 2000). The earliest electron micrograph of F₀F₁ showed only a central stalk (Kagawa and Racker, 1966b). This single stalk F₀F₁ was reconstituted from both ³H-acetyl-F₁ and urea-cholate-treated F₀ devoid of the peripheral stalk (Kagawa and Racker, 1966b) (Fig. 1, upper right). This single-stalk F₀F₁ restored coupling activity by adding OSCP and other extracted crude factors (Kagawa and Racker, 1971) by forming the stator. In fact, X-ray crystallography of detergent-treated F₀F₁ revealed a single stalk (Stock *et al.*, 1999), because both F₀-a and F₀-b were lost. The resulting single-stalk F₀F₁ shows the rotation of F₀ that reflects the $2\pi/3$ steps of γ , not the $\pi/5$ steps of F₀-c subunits (Tsunoda *et al.*, 2000).

PRIMARY STRUCTURE OF THE α AND β SUBUNITS

We sequenced TF₀F₁ (Kagawa *et al.*, 1986) and human MF₁ (Ohta and Kagawa, 1986). The numbers of amino acid residues in the aligned sequences in T α and T β are shown in Fig. 2. With respect to nomenclature for the amino acid residues in the subunits α , β , γ , etc., of TF₁, MF₁, and EF₁, "T β K35," for example, means that a lysine (K) residue is located at No 35 in the β subunit of TF₁ (T β). The equivalent residues in different F₁s are expressed as T β K164 (=M β K162, E β K155). A mutation is abbreviated as "T β K164I," which means "the residue K164 of T β is changed into I164." The primary structures of the α and β subunits from TF₁ (T α and T β) and MF₁ (M α and M β) are homologous with 59% sequence identity between T α /M α and 68% between T β /M β . Conserved residues (green and blue letters in Fig. 2) among TF₁, MF₁, and EF₁ are closely related to catalytic and regulatory functions. Species-specific residues (black letters in Fig. 2) may have phylogenic and ontogenic meaning. A detailed phylogenic tree of F₁ was drawn using the protein database. According to our studies using transgenic mice, γ -subunit isoforms are produced by splicing during ontogenesis (Ichida *et al.*, 2000; Kagawa *et al.*, 1997).

DOMAINS OF α AND β , AND OPEN-CLOSE MOVEMENTS

Three Domains of the α and β Subunits

The tertiary structure of T α , T β (Shirakihara *et al.*, 1997), M α , and M β (Abrahams *et al.*, 1994) consists of

three domains (Fig. 3). These are an N-terminal β -barrel (T α 21–94, T β 1–82), a central nucleotide-binding region (T α 95–371, T β 83–354), and a C-terminal α -helical bundle (T α 375–502, T β 355–473) (Figs. 2 and 3). The barrel domain contains six β sheets (a–f) arranged like an IgCI fold (pairs of a–b, c–f, and d–e strands) and the barrel structure is found in SH3 (*src* homology 3). The nucleotide-binding domain consists of a nine-stranded β sheet (1–9) surrounded by eight α helices (A–H) and a small additional antiparallel β sheet (0 and 2.1) (Fig. 2) structures, which are similar to the nucleotide-binding domains of G proteins. The C-terminal domain consists of six α -helices in M α and T α (helices 1, 2, 4, 6–8), and in M β and T β (helices 1–6) (Fig. 2). Thus, the largest difference between α and β is in the C-terminal region. The α -helical bundle is also found in helical cytokines. The loops are located at the central cavity and peripheral surface and play important roles.

THE OPEN- (β_E) AND CLOSED FORM (β_D , β_T) OF β SUBUNIT

The nucleotide-free crystal of $\alpha_3\beta_3$ (Shirakihara *et al.*, 1997) shows strict threefold symmetry. In contrast, because of the presence of one copy of $\gamma\delta\varepsilon$ and $\alpha_3\beta_3$ per F₁ molecule, the threefold symmetry in the crystallography of F₁ is an approximation and γ is disordered (Bianchet *et al.*, 1998). Bovine heart MF₁ crystallized in D₂O containing AMP-PNP, ADP, Mg²⁺, azide, and Tris shows three different states of β (Abrahams *et al.*, 1994) that have ADP (β_D), AMP-PNP (β_T), and no nucleotide (β_E), respectively. The original rotational theory assumed the loose-, tight-, and open- β conformations (Boyer, 1997). The β_T and β_D have very similar conformations, but β_E is quite different (Abrahams *et al.*, 1994), and both β_D and β_T are classified as closed forms (C), while β_E has the open form (O). T β_E in $\alpha_3\beta_3$ superimposes well on the whole M β_E . Thus, $\alpha_3\beta_3$ is OOO type (Shirakihara *et al.*, 1997) and bovine MF₁ crystal is CCO type (Abrahams *et al.*, 1994). However, rat liver MF₁ crystallized in H₂O containing ATP and P_i (without Mg²⁺) has threefold symmetry and the structure seems to be of the CCC type (Bianchet *et al.*, 1998), because all M β s have ADP and two may have P_i. However, formation of the CCC type by adding ADP to $\alpha_3\beta_3$ causes its dissociation into $\alpha_1\beta_1$ (Sato *et al.*, 1995; Kagawa *et al.*, 1992), because two β_D s collide at T β 1396 (Fig. 3, left bottom I in red circle). In fact, the β - β is cross linked at these residues and the CCC type was refuted (Tsunoda *et al.*, 1999).

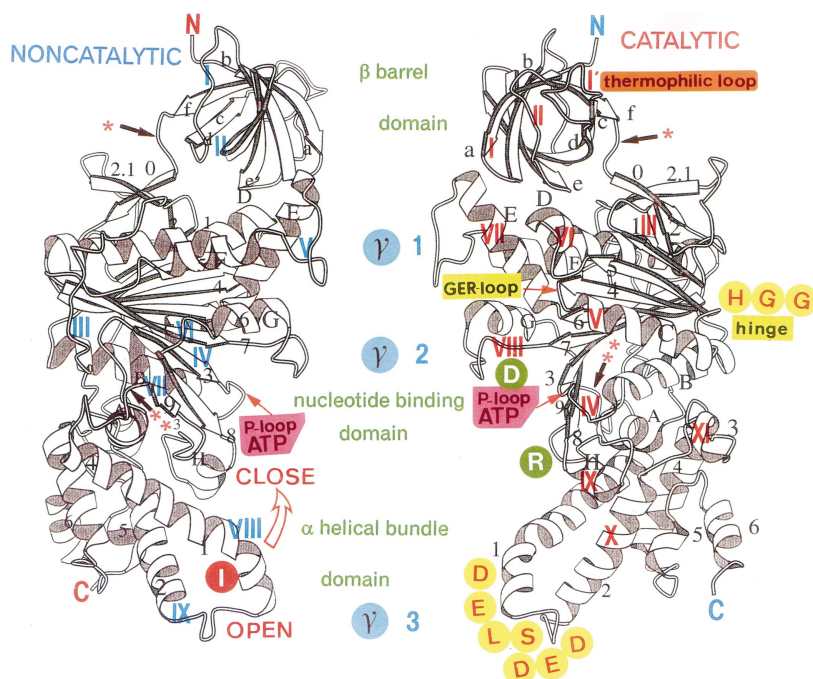


Fig. 3. Three-dimensional structures of T β E. Threefold axis is vertical, so that views are toward the α/β subunit interfaces. Left, noncatalytic interface (blue I–IX indicates contact areas). Right, catalytic interface (red I–XI indicates contact areas). Green letters, domain names and the border of domains are marked by red asterisks. D and R in green circles are T β D331 and T β R333, respectively, at the entrance to the crevice of the P-loop. I in red circle is T β I386 of $\beta\beta$ contact. H and Gs in yellow circles are the hinge residues. (Modified from Shirakihara *et al.*, 1997.)

THE CATALYTIC AND NONCATALYTIC α/β INTERFACES

The α/β Interfaces

The α and β contact each other via catalytic and noncatalytic interfaces, and these are expressed as primary structures in Fig. 2 for T α and T β , and Table I for M β . Except for the thermophilic loop, homological contact areas (I–XI) are located in both β and α at catalytic (red and pink bars in Fig. 2), and noncatalytic (blue and green bars in Fig. 2) interfaces. These areas in T β are shown in Fig. 3.

The barrel domain harbors the universal contact areas I and II (Table I and Fig. 2). The common electrostatic bond in II is T β R72–T α E67 (=M β R71–M α E67). At the catalytic nucleotide-binding domain, areas III, V, VI, VII, and VIII are universally detected (Table I). However in T β E, the P-loop contact area IV is latent, in contrast to that area in T α E (Fig. 2), M β T, and M β D (Table I). M α R373 interacts with oxygens of β - and γ - P of ATP bound at IV. In V, the common electrostatic bonds are T β R193–T α D339 and M β R191–M α D347. In VI and VII of β T and β D, M α F299–M β M222 and M α S344–M β R260, re-

spectively, interact. However, we identified no direct contact in the α -helical bundle domain in T β E of OOO type F₁. In the catalytic α/β interface of M β E, the contact areas (17.6 nm²) are homologous to those of T β (Table I). The areas of M β D (30.3 nm²) and M β T (22.0 nm²) are increased to 11 and 10, respectively (Table I). This is caused by the $\pi/6$ upward motion of the C-terminal domains.

The seven main contact areas of the noncatalytic α/β interface of M β E in are summarized in Table I (areas I–VII). In M β E, the contact areas of the noncatalytic α/β interface are homologous to those of T β E (Fig. 2). The noncatalytic site of α accepts M β D359 (VII) that interacts with the ribose of AMP–PNP. In the closed form of M β , these areas are changed in M β D (9) and M β T (7) (Table I, bottom).

The catalytic interface of α has five contact areas between β E, but the number of these areas increased to 8 and 7 between β D and β T, respectively. In the closed M β , α R373 interacts with, in addition to AT(D)P, both V (β R189 β R191) and IV (β G159), because α -helices B and C move closer. However, the noncatalytic interface of α has seven contact areas between β E, β D, and β T. The target α areas of I–XI of β are shown in Table I. Because

Table I. Catalytic and Noncatalytic $\alpha\beta$ Interfaces of MF₁ β Subunit

Ligand ^a	Catalytic $\alpha\beta$ interface (I–XI: contact areas and their location)																			
	The major charged groups for intersubunit interactions: D, E, R, H, and K																			
	I Sheet <i>a</i>		II Loop <i>e–f</i>				III Sheet 1, 2			IV P-loop		V Helix C			VI Sheet 5, helix D					
	14	18	64	66	68	71	94	102	104	158	159	189	192	194	197	219	222	226	229	
E	VAVIG		DGT–GLVR				I–IDE			—		RTRE–ND–Y			Y–MNEP–R					
D	VAVI		D–TEGLVR				ID			AG		RTR–ND–Y			Y–MNEPP–R					
T	AVI		D–TEGLVR				I–IDE			AG		RTR–ND–Y			Y–MNEP–R					
target α	sheet <i>d</i>		sheet <i>c</i> , loop A				loop A			helix G		helices F, G			loop A, helix F					
Catalytic $\alpha\beta$ interface (I–XI: contact areas and their location)																				
	VII Helix E, γ -Loop F					VIII γ -Loop G			IX α -Helix H			X α -Helix 2			XI α -Helix 3, 5					
	260	263	267	273	281	311	314	319	337	341	345	388	404	412	423	424	453	455	458	
E	–E–AL–G–					—A—			—			—			—					
D	R–Q–E–A—V–Y					Y–AD–D			R–ELG–Y			I–V–R–R			VFTGH–P–QY					
T	R–Q–E–A–P–V					Y–ADD–D			R—			—			VF—Q					
target α	loop F, helix F, γ					helix G, γ			loop G, ATP–Y			loop 2, γ			loop 1–helix 1					
Noncatalytic $\alpha\beta$ interface (I–IX: contact areas and their location)																				
	I Loop <i>bc</i>			II Sheet <i>d</i>			III Loop <i>d–A</i>				IV Sheet 3		V γ -Loop F, helix F							
	29	31	32	51	54	56	119	124	128	130	151	273	275	278	283	287	290	294		
E	L–PI			QHLGES			EAPEFV–M–VEQ				K		GRIPSA–PT—T–G—E							
D	L–PI			QHLGES			EAPEFV–M–V–Q				K		GRI–SA–PT—T–GT–E							
T	L–I			QHLGES			EAPEFV–MSV–Q				—		GRIPSA–PT–AT–GT–E							
target α	loop <i>f</i>			sheet <i>a</i> loop <i>f</i>			sheet <i>f</i> , helix C				loop C		helix F, loop F							
Noncatalytic $\alpha\beta$ interface (I–IX: contact areas and their location)																				
	VI γ -Loop, helix G			VII Sheet 9				VIII Helix 1					IX DELSEED-loop							
	317	316	330	351	355	356	359	368	372	375	380	383	387	395	397	400				
E	–TA–FAH–D			—R—D				—					—							
D	LT–FAHLD			LDSTR—D				—QK—QD–SL–I					ELS—D							
T	–T–FAHLD			LD–TR—D				Y–R–QK—QD–S—					—							
target α	loop B, sheet 4			P-loop, helix C				helix H, loop–helix 2					loop–helix 2							

^a Ligands on the β subunits: E, no ligand; D, ADP; T, AMP–PNP.

of the high homology between TF₁ and MF₁, the total numbers of atoms in the areas of catalytic interface of T β _E and M β _E were 91 and 93, respectively, if the TF₁-specific area of the thermophilic loop (11 atoms, I' in Fig. 2) is removed. Those of the noncatalytic interface of T β _E and M β _E were 112 and 114, respectively. In contrast, the atom numbers (in parentheses) in the contact area were significantly increased in the closed forms: catalytic interface of M β _D (176), M β _T (131), and noncatalytic interface of M β _D (141) and M β _T (139), mainly at the lower half of the β .

The Immobile Areas for the Stator

The conformations of bovine M α loaded with Mg-AMP–PNP (Abrahams *et al.*, 1994), rat liver M α loaded with Mg-ATP (Bianchet *et al.*, 1998), and T α without ligands (Shirakihara *et al.*, 1997) are equally open. The entire structure of the β -barrel domain of β _E superimposes well on that of β _D and β _T. Torsional strain in the β -barrel domain is low because of the distance from the β - γ contact points, slender mainchain connection (Fig. 3, red*), and firm binding through I, I', and II. Thus, the β -barrel domain

of both α and β is immobile during the rotation. The stator may be connected to $\alpha_3\beta_3$ via δ attached to the barrel domain. In fact, OSCP was cross linked to the N-terminal of α (Xu *et al.*, 2000).

REFOLDING OF TF₁ SUBUNITS AND ELECTROSTATIC FORCE

In the stress-strain analysis, F₁ is not a uniform rubber ball, but the assembly of heterogeneous solid composed of α -helices, β sheets, and coils (Figs. 1 and 3). The forces to form the tertiary structure are propensity to form α -helices and β sheets, as well as external hydrophilicity and internal hydrophobicity of $\alpha\beta$. TF₁ was completely denatured and refolded (Yoshida *et al.*, 1975). During subunit refolding, the molecule assumes an expanded state called the “molten globule phase” in the order of milliseconds or less (Arai and Kuwajima, 2000). CD (Yoshida *et al.*, 1979), and D–H exchange (Ohta *et al.*, 1980) confirmed the refolding (Kagawa, 1999) (Fig. 1, upper left). In contrast to H bonding for secondary structures, etc., and van der Waals force for helical bundle formation, etc., electrostatic force is long-reaching and plays an important role in domain movements (Hall and Pravitt, 1984). The primary structures of the TF₁ subunits contain refolding information which differs from those of other F₁. Stability (Argos *et al.*, 1979) and folding (Jaenicke and Boehm, 1998) of thermophilic proteins can be achieved by adding many small residue substitutions to the corresponding mesophilic proteins without significant change in backbone conformation. The rule for the general thermophilic protein (Argos *et al.*, 1979) was confirmed not only in TF₁ (Kagawa *et al.*, 1986) but also in hyperthermophilic V-ATPase (Shibui *et al.*, 1997), that is homologs of F_oF₁. The residues mainly contributing to the increased propensity of the secondary structure in T β are located near the barrel domain: sheets a (V6, I7, and Q8), c (K32 and H34), o (S84 and V85), and l (V95), and exceptionally in helix l (E361, Q365, K369, E375, and E379). The highly conserved α -helices (A–G and 2–6) contain residues with almost equal propensity in both MF₁ and TF₁.

The basic mechanism of the synthesis/hydrolysis of ATP is equal in TF₁ and mesophilic F₁s. Thus, the different stability is mainly limited in the immobile areas I, I', and II (and helix I) not to areas III to XI.

ASSEMBLY MUTANTS

Site-directed mutagenesis produces many defective subunits (Futai *et al.*, 1989; Yagi *et al.*, 1999; Omote *et al.*, 1995). We located these mutant residues in X-ray crystallography of β_E , β_D , and β_T . If “assembly mutant

residues” are strictly defined as those localized at the $\alpha\beta$ -contact area to bind both subunits, very few are present: E α E299 (=M α E299, IV of α), E α S347 (=M α S344, V of α), E β E185 (=M β E192, area V of β), E β 192 in helix C (V), and E β G214 (=M β G227, VI of β) of catalytic interface (Table I). Any denaturing mutation of mesophilic F₁ may be false “assembly mutation.” An example of a real assembly mutant is T β Y277F, which cannot assemble (Yagi *et al.*, 1999) at the noncatalytic interface V, and this tyrosine is conserved among species. Functionally, residues, such as E β E185, can be involved in cooperative catalysis (Omote *et al.*, 1995).

Other defective residues may indirectly affect conformations around the interfaces and the P-loop. For example, the E β G149C mutation in the P-loop is suppressed by E β S174F in strand 4, which is quite far from the P-loop (Futai and Omote, 1996). Furthermore, this defective coupling activity of the E β S174F mutant was suppressed by the second mutation, E α R296C, in helix F of α across the α/β interface. However, both mutations are not in direct contact. The fact that a mutation in β is suppressed by mutation in α , supports the notion of indirect $\alpha\beta$ interaction. If this kind of mutation causing indirect interactions is included in the catalytic mutation, the distinction between catalytic and assembly mutation becomes blurred.

In vivo Refolding and Assembly of F₁ Require Chaperones

The primary folding and assembly of peptides synthesized by mRNA translation require chaperones such as Atp11p protein in MF₁ (Wang and Ackerman, 2000). Atp11p binds to a region of the nucleotide-binding domain of M β located between M β G114 and M β L318 (IV–VIII) and assembled M α . We cloned and characterized a hyperthermophilic chaperone to use as a tool for the refolding of subunits (Yoshida *et al.*, 1997). Since, *in vitro* refolding and reassembly are difficult in mesophilic F₁s, the *in vivo* assembly mutant may include mutants that are defective in the chaperone step.

NUCLEOTIDE-BINDING LOOPS

P-Loop in Area IV

The P-loop is found in many nucleotide-binding proteins, including α (noncatalytic site) and β (catalytic site) of F₁ (Abrahams *et al.*, 1994). The catalytic P-loop (GGAGVGKT; T β 158–165, Fig. 3, pink) is located between sheet 3 and helix B. Most amide groups in the P-loop are involved in H bonding to the β -phosphate of AT(D)P. Thus, binding of ADP and ATP can equally form closed β . T β K164 (=M β K162) in the P-loop binds to the

γ P of ATP, and T β K164I lost both nucleotide binding and catalysis abilities (Yohda *et al.*, 1988). The true substrate bound at the P-loop is Δ , $\beta\gamma$ -bidentate Mg, ATP complex (Senter *et al.*, 1983). Mg²⁺ is required for the expression of catalytic site-binding cooperativity and high-affinity binding at catalytic site 1. The K_d values for sites 1, 2, and 3 are about <1nM, 1 μ M, and 100 μ M, respectively, in EF₁ (Weber and Senior, 2000). The hydroxyoxygen of T β T165 in the P-loop directly coordinates with Mg²⁺. The ESR of both MF₁ and TF₁ spin labeled with 2-azido 2', 3'-SL-ATP revealed the tightening role of Mg²⁺ (Burgard *et al.*, 1994).

GER Loop in Area V

The nucleotide-binding domain of β accommodates the conserved T β E190 (=M β E188, E β E181) that may function as a general base to activate bound water by abstracting a proton. This residue is localized at loop 4C (=GER loop), and interacts with the γ -phosphate of ATP via a water molecule (Abrahams *et al.*, 1994). The direct contact of helix C at V of the catalytic interface (Fig. 2) and the hinge motion of the C-terminal domain containing DELSEED in M β E, rotates away from the core of the particle by almost $\pi/6$ (Abrahams *et al.*, 1994). The resulting widening of the P-loop-T β E190 distance causes the release of Mg-ATP from the catalytic site.

Loop 7G in Area VIII

Loop 7G is located between sheet 7 and helix G of β . This loop is close to the P-loop in the closed form, but is distant from the P-loop in the open form. The conformational change is transferred to the P-loop via loop 7G, a putative switch region (Noji *et al.*, 1996) located in VIII (Fig. 2) near T β D315, the closest point to the γ -axis (Fig. 3, γ 2). G proteins and myosin (Sasaki *et al.*, 1998) contain switch domains 1 and 2, and the latter is, to some extent, similar to VIII. As in G proteins, a nucleotide is bound around the α -helix B and the P-loop. The alanine scanning mutagenesis (D454A, G457A, F458A, and E549A) of myosin removed the motor function and G457A lost ATPase by blocking main-chain rotation because G457 corresponds to the γ -phosphate sensor glycine (G60 of Ras) in the G-protein (Sasaki *et al.*, 1998).

PROPERTIES OF ISOLATED SUBUNITS

Monomeric Structure of the α and β Subunits

The diffraction pattern of mutant (Y341L) T β crystallized in the absence of Mg-nucleotide shows unit-cell dimensions of $a = 232\text{\AA}$, $b = 66\text{\AA}$, and $c = 80\text{\AA}$

(Saika *et al.*, 1994). The whole structure superimposes well on that of M β E (Abrahams *et al.*, 1994) and of T β E (Shirakihara *et al.*, 1997). This result indicates that the "open form" of β is not caused by the $\alpha\beta\gamma$ interaction (Fig. 1 lower left). The conformation of isolated T α and T β was changed by adding Mg-AT(D)P as shown by IR (Ohta *et al.*, 1980), NMR (Tozawa *et al.*, 1995), CD, neutron scattering, and X-rays (Kagawa, 1999). In fact, the cross linking of T β D311 and T β R333 (Fig. 3, green D and R) fix the isolated β in a closed form, in which α -helix C (V) and α -helix B (IV) become parallel (Ren *et al.*, 1999). The following X-ray data constitutes a personal communication from Prof. K. Miki. Because of the absence of the surrounding subunits, some contact areas, including the thermophilic loop (I' in Fig. 2) of T β , are disordered. Among the contact areas at the catalytic sites in the T β shown in Fig. 2, the structures of areas I, II, III, VI, and VII are located very close to M β E and T β E. However, V in α -helix C (the closest helix to α) has no adjacent α -helices F and G in T α , and part of α -helix C is disordered and expanded to the T α side. Thus, both side-chain and main-chain folding of T β E190 and T β R191 in the GER loop of the isolated T β shifted from their positions in active $\alpha 3\beta 3$. The C-terminal regions are disordered, and the temperature factor (80\AA^2) was much larger than that in the barrel- and nucleotide-binding domains. Among the noncatalytic contact areas, the structures of areas I and III have no interacting α -helices, and slide to the direction of missing α -helices. Except for the disordered VI, the other areas are almost identical to β E in the oligomer.

Conformational dynamics of the isolated T β were monitored by T β H179 at the conical tunnel entrance and those of T β H200 were examined at catalytic V using [¹H]NMR. Binding of Mg-AMP-PNP significantly affected the chemical shift of both residues, indicating a ligand-induced conformational change (Tozawa *et al.*, 1995). The 3,5-proton resonance of 12 tyrosine residues in T β revealed a Mg-ATP-dependent signal shift in Y199 at V, in Y307 at VIII, etc. (Yagi *et al.*, 1999).

THE $\alpha_1\beta_1$ PROTOMER AND CATALYTIC INTERFACE

The assembly of $\alpha_1\beta_1$ protomer forms a catalytic interface (Kagawa *et al.*, 1992). A sequence specific to TF₁ is located in the β barrel (Figs. 2 and 3, I' T β 36–42, or thermophilic loop). The thermophilic loop holds T α and T β tightly at the catalytic interface by an H bond (β D40– α R90) and by van der Waals interactions (β N38– α S21 and β E41– α M48).

The nonactive catalytic site in the T β monomer is converted into the active catalytic site by a supply of residues in T α (T α T338 in VI and T α R365 in VII). The

guanidium of M β R373 and the ϵ -ammonium of M β K162 interacts with the γ -P of Mg-ATP (Abrahams *et al.*, 1994). T α R365 (=M α R373) may stabilize the terminal phosphate in a pentacoordinate transition state of the Δ, β, γ -bidentate Mg-ATP. However, the mutant of this residue still show unisite ATPase activity and E α R376 (=M α R373, T α R365) may be required for the promotion of the steady state (Le *et al.*, 2000). M α R373 interacts with G159 (IV), R189 and R191 (V), and F424 (XI) in addition to the nucleotide (Table I). The kinetics of the ATPase of $\alpha_1\beta_1$ were typical Michaelis-Menten type (Saika and Yoshida, 1995), with only one K_m ATP value of 70 μ M, and a V_{max} value of 0.1 unit/mg, without cooperativity. The H bond between T β E201 (=M β E199) and T β T165 (=M β T163) stabilizes the open form, as shown by an experiment that used $\alpha_3\beta_3\gamma$ containing the mutant T β E201C (Ren and Allison, 2000). T β T165 interacts directly with Mg coordinated to ADP and T β becomes closed form.

THE $\alpha_3\beta_3$ OLIGOMER AND NONCATALYTIC INTERFACE

The noncatalytic, in addition to the catalytic interface, is now formed by the assembly of three $\alpha_1\beta_1$ into $\alpha_3\beta_3$ (Kagawa *et al.*, 1992; Sato *et al.*, 1995). The noncatalytic nucleotide-binding site is formed at the α P-loop with T β D326, T β R352, and T β Y364 on the noncatalytic α/β interface. In contrast to $\alpha_1\beta_1$, the ATPase activity of $\alpha_3\beta_3$ is cooperative. Depending on the occupancy of the catalytic sites by an increasing Mg-ATP concentration [Mg-ATP], the ATPase activities of F₁ consist of uni-, bi-, and trisite types. Unisite catalysis is measured at the substoichiometric Mg-ATP concentration ([Mg-ATP] < [F₁]). The unisite activity is very low, and the apparent K_m ATP value is below 20 nM. The apparent K_m ATP values of oligomeric ATPase of $\alpha_3\beta_3$ were about 150 (bisite) and 490 μ M (trisite). The K_m ATP values of the rotational ATPase of TF₁ (similar to $\alpha_3\beta_3\gamma$) were about 80 (bisite) and 490 μ M (trisite). Only one mole of [³H]Bz-ADP per $\alpha_3\beta_3$ inhibited its ATPase activity, as both were MF₁ and TF₁ (Aloise *et al.*, 1991). This single-hit inactivation and cooperativity are inherent properties of the symmetrical $\alpha_3\beta_3$, but are not due to the interaction with γ .

THE $\alpha_3\beta_3\gamma$ AND ITS HYBRIDS WITH DEFECTIVE α AND β

The major role of γ , a rotating “eccentric cam” in $\alpha_3\beta_3$, is to release Mg-ATP from β_T by changing the conformation (Leslie and Walker, 2000; Oster and Wang,

2000). Because of negative cooperativity between the catalytic sites, the first ATP binds very tightly ($K_d = 10^{-12}$ M), and the last hardly binds at all. Because of this extreme K_d value, ATP is synthesized as F₁-ATP complex without the addition of external energy. In fact, incubating ADP-TF₁ (1:1) complex with P_i and Mg²⁺, produces TF₁-bound ATP (ATP: ADP: TF₁ = 0.55:0.45:1) (Yohda *et al.*, 1986). The addition of γ to $\alpha_3\beta_3$ also changes sensitivity to an inhibitor (Paik *et al.*, 1993). Two schemes of hydrolytic cycles have been proposed (Leslie and Walker, 2000). In scheme (a), ATP binding to β_E results in a major conformational change in β_E , which, in turn, rotates γ by $2\pi/3$, and converts β_D to an open conformation. In the alternative scheme (b), ATP binding to β_E promotes ATP hydrolysis at β_D , and the hydrolysis results in a large conformational change in β_D and the release of ADP + P_i. Scheme (a) is more probable, because large conformational change in the isolated β_E was revealed by NMR (Yagi *et al.*, 1999). Another new mechanism of ATP hydrolysis via CCC type (full occupancy of three sites) has been proposed (Weber and Senior, 2000). This is based on the nucleotide-dependent fluorescence change of E β Y331W without X-ray crystallography. As discussed in the previous section, only modified CCC' type is compatible with the results on TF₁ (Kagawa, 1999; Tsunoda *et al.*, 1999). There should be a transient β_E to allow nucleotide release and binding.

Hybrid $\alpha_3\beta_3\gamma$ Containing 1–3 Subunits with Defective P-Loop

The subunit–subunit interactions via $\alpha\beta$ interfaces were analyzed using homogeneous hybrids of T $\alpha_3\beta_3\gamma$ containing 1, 2, or 3 defective subunits (αi or βi). To isolate T $\alpha_3\beta_3\gamma$ containing 1, 2, or 3 αi or βi , a decaglutamate tag was introduced to αi (Bald *et al.*, 1999) or βi (Amano *et al.*, 1996) (Fig. 1, bottom, green). After reassembly of the wild type and the mutant into the mixture of four hybrids, they were separated into homogeneous hybrids by ion-exchange chromatography.

The T $\alpha_3\beta_3\gamma$ hybrids with T β carrying an inactive catalytic site ($\beta i =$ T β E190Q tagged with decaglutamate) revealed that normal steady-state ATPase activity, catalytic cooperativity, and unisite ATPase activity require(s) 3, 2, and 1 intact β (s), respectively (Amano *et al.*, 1996). In α , T α QQ200 (M α Q208) of the noncatalytic site occupies a position equivalent to that of catalytic T β E190. Thus, α , like the T β E190Q mutation cannot hydrolyze ATP.

The hybrid T $\alpha_3\beta_3\gamma$ containing 1, 2, or 3 defective noncatalytic site(s) in the α subunits (αi) were prepared. To destroy the noncatalytic site of α , the following mutations

were introduced: $\alpha_i = T\alpha K175A/T176A/D261A/D262A$ with a decaglutamate tag (Bald *et al.*, 1998). The hybrid $T\alpha_3\beta_3\gamma$ displayed a substantial steady-state activity depending on the number of α_i (Fig. 2. of ref. Amano *et al.*, 1999 with different protein concentrations). The result indicates that one wild-type α is sufficient to maintain the steady-state catalysis and can potentially relieve all three β s from inhibition by Mg-ADP. $T\alpha D261N$ (Yohda *et al.*, 1988) failed to dissociate inhibitory Mg-ADP (Jault *et al.*, 1995). These results also suggest that the interaction between three noncatalytic sites is not so powerful as that between catalytic sites, which are strictly required for steady-state ATPase. Moreover, $TF_0-(\alpha_i)_3\beta_3\gamma$ reconstituted into liposomes can synthesize ATP and is insensitive to azide (Bald *et al.*, 1998). Thus, the noncatalytic site is not necessary during ATP synthesis.

INTRAMOLECULAR ROTATION

The single molecule analysis of rotating F_0F_1 (Sambongi *et al.*, 1999) is controversial (Tsunoda *et al.*, 2000). Thus, we discuss here only the real time imaging of $\alpha_3\beta_3\gamma$ fixed on a bead (Kinoshita, 1999). In the presence of Mg-ATP, the filament attached to γ rotated in a counterclockwise direction ($\beta_D \rightarrow \beta_E \rightarrow \beta_T \rightarrow$) when viewed from the membrane side. The work completed by the rotation is the frictional torque multiplied by the angle of rotation. The hydrodynamic frictional drag coefficient (ξ) of the actin filament for the propeller rotation is given by $\xi = (\pi/3)\eta L^3 / [\ln(L/2r) - 0.447]$, where η (10^{-3}Nsm^{-2}) is the viscosity of the medium, L the length of the actin filament (1–4 μm), and r (5 nm) the radius of the filament (Yasuda *et al.*, 1998). The observed rate of filament rotation rates at 2 mM ATP are 7, 1, and 0.1 rps (revolutions per second), when the lengths of f-actin are 1, 2, and 4 μm , respectively (Yasuda *et al.*, 1998). By shortening the f-actin and by extrapolating the external work to zero, the calculated internal friction, if any, was very low. The frictional torque $\xi\omega$ was about 40 pN/nm, where ω is the angular velocity. The stoichiometry of three ATPase catalytic sites per single γ in F_1 indicates a nonload rotary rate of 17 rps as calculated from the ATPase activity of 52 ATP molecules metabolized per second. The ω of γ in the $\alpha_3\beta_3$ decreases at ATP concentrations below 2 μM (Yasuda *et al.*, 1998). For short f-actin (0.8–1.2 μm), the rate ω at 20 nM–2 mM ATP fitted simple Michaelis-Menten kinetics [$V_{\max}(\omega) = 3.9$ rps, and $K_m(\omega) = 0.8 \mu\text{M}$] (Yasuda *et al.*, 1998). Thus, the rotation may be fueled by individual ATP molecules without requiring the simultaneous consumption of two or more of them. The rotational rate depends on both the frictional load and ATP binding. At

higher ATP concentrations (tristite catalysis), the rotational rate was saturated at the level determined by the frictional load.

CROSS LINKING OF SUBUNITS TO FIX THE DOMAIN MOTION

The α - β Cross Linking

Both the catalytic and noncatalytic sites of F_1 were cross linked with 2,8-diN3-ATP (α and β , 9.6 Å apart), to form both $\alpha\beta$ dimers and trimers (Schäfer *et al.*, 1995). The adenine-binding pocket cross linked with 2-azido-ATP is $T\beta Y341 (=M\beta Y345)$, and the mutants $T\beta Y341A$ (or C or L) have low activity (Kaibara *et al.*, 1996).

The β - β Cross Linking of Closed Form β s

The presence of the two closed β and one open β structure CCO of F_1 in the reaction was confirmed by cross linking two closed β at the position of $M\beta I390 (=T\beta I394)$, Fig. 3, I in red circle) of each β (Tsunoda *et al.*, 1999). The distance between the nearest α carbon atoms of the $M\beta I390$ in $M\beta_D$ and $M\beta_T$ was only 0.79 nm, while those between $M\beta_E$ and $M\beta_D$ and $M\beta_E$ and $M\beta_T$ were 2.58 and 2.40 nm, respectively. The distance between $T\beta I386$ of $T\beta_E$ in $\alpha_3\beta_3$ is 3.27 nm (Shirahihara *et al.*, 1997). Thus, the β - β cross link could form in the CCO, but not in COO and OOO types of TF_1 . The mutant $T\beta I386C$ was assembled into the $T\alpha_3(T\beta I386C)_3\gamma$ subcomplex. $T\beta_T$ and $T\beta_D$ were cross-linked via a disulfide bond between the two $T\beta I386C$ by oxidation in the presence of Mg-AT(D)P (Tsunoda *et al.*, 1999). The β - β cross-linked $T\alpha_3T\beta(T\beta I386C)_2\gamma$ was inactive, but was reactivated by reduction with dithiothreitol. The CCC-type MF_1 (Bianchet *et al.*, 1998) is impossible because of steric collision.

At the entrance to the crevice of the P-loop, the distances between $M\beta D315 (=T\beta D311)$, in VIII) and $M\beta R337 (=T\beta R333)$ in IX) are 0.30, 0.37, and 1.03 nm in $M\beta_T$, $M\beta_D$, and $M\beta_E$, respectively (Fig. 3, D and R in green circle). Cross linking of $T\beta D311C$ and $T\beta R333C$ fix β in the closed form in which α -helix C and α -helix B become parallel (Ren *et al.*, 1999). In the reassembled $\alpha_3(\beta D311C/R333C)_3\gamma$, less than two $\beta D311C/R333Cs$ were cross linked in the presence of Mg-nucleotide, thus, the CCO or COO types can arise. However, under high nucleotide conditions, all β s bind the ligand (Weber and Senior, 2000), and an approximate threefold symmetry has been detected in rat liver MF_1 (Bianchet *et al.*, 1998). As

will be discussed below, the β DELSEED region is elastic (Kagawa, 1999), and C'CC form may arise, in which C' is deformed C.

The β - γ Cross Linking

Since γ is the rotational axis, any cross linking between γ and $\alpha_3\beta_3$ results in a loss of multisite catalysis, but unisite catalysis remains in the $\beta\gamma$ cross-linked EF₁ (Garcia and Capaldi, 1998). Three points of contact between M γ and M β are catalytic VII and VIII (D316, T318, and D318), and noncatalytic IX. These areas of α/β interfaces, are directly deformed by γ during rotation. In β_T , the noncatalytic IX is the closest to γ RGL (γ 3 in Fig. 3), which is 2 nm from the central axis may transfer torque to DELSEED. The conserved M γ Q255 (=E γ Q269) may also transfer torque via M β D316 and M β D318 in VIII (γ 2 in Fig. 3) (Omote *et al.*, 1998). The interaction between γ 1 and VII is not established.

THE HINGE OF DOMAIN MOTIONS

F₁ is a flexible entity (Hayward, 1999; Yan *et al.*, 1999), and it is necessary to detect the hinge point between open and closed β that may generate the torque of γ rotation (Oster and Wang, 2000; Kagawa, 1999). We cannot estimate elasticity solely on the physical property of the separate α -helix and β sheet. These structures are mechanically grouped in a protein. For example, in β_T , α -helix G, β sheets 7, 3, and 8, and α -helix B are packed in parallel, but in β_E , the distances between the helix-sheets-helix, especially that between β sheet 7 and 3, are widened. Thus, hinge of the domain is important to estimate elasticity. An angle formed by four consecutive C α atoms in polypeptide backbones was computed from X-ray crystallography of the proteins. The difference in this angle from the equivalent residues between the open and closed forms was used to detect the pivot (Yan *et al.*, 1999). The pivot residues of T β that change the ϕ and ψ angles of a Ramachandran plot by more than $5\pi/9$ on addition of AT(D)P are V162, G163, H179, G180, and G181 (Abrahams *et al.*, 1994). V162 and G163 belong to the P-loop (ATP-binding site) at IV. H179, G180, and G181 are located at the loop between the loop B-4 located at the outer end of β (Shirakihara *et al.*, 1997) (Fig. 3, H,G,G in yellow circle). The open-close angle between M β DELSEED (or T β DELSDED) and VIII of the catalytic interface (loop 7G) is about $\pi/6$. The loop B-4 connects the P-loop and the GER loop near V in the catalytic interface (Fig. 2). The residues (T β G180, T β E361, T β S30, and T β H179) in the

rotation also show large changes in the ϕ and ψ angles (Masaie *et al.*, 2000). The simultaneous mutagenesis of T β H179A, T β G180A, and T β G181A caused almost a complete loss (99%) of ATPase activity. The steady-state ATPase activities of $\alpha_3(\text{mutant } \beta)_3\gamma$ were 65, 23, 4.3, and 160 s⁻¹ in wild-type T β , T β H179A, T β G180A, and T β G181A, respectively. However, T β G180A enhanced and T β G181A abolished the propensity to generate the kinetically trapped Mg-ATP inhibited form (Masaie *et al.*, 2000). Despite the hindrance to the formation of T β_E in the $\alpha_3(\text{mutant } \beta)_3\gamma$ thus generated, the mean rotational torque was not changed (Masaie *et al.*, 2000). Although, other causes may explain the loss of ATPase activity with normal torque in the hinge mutant, the rotational velocity of the mutants is only 0.2 to 5 (0.6 to 15 ATP/sec) compared with 4.3 to 160 ATP/sec hydrolyzed (turnover rate) in free $\alpha\beta\gamma$. Thus, only a few hits of ATP at the catalytic site are required to drive rotation, even when ATPase levels are dramatically decreased.

ELASTIC ENERGY STORED DURING THE τ_{step}

The time per $2\pi/3$ revolution (τ_{120}) is composed of the cycle time for unloaded ATPase, (τ_{ATP}) and $\tau_{\text{step}} = (2\pi/3)\xi/N$, thus, $\tau_{120} = \tau_{\text{ATP}} + \tau_{\text{step}}$. The τ_{step} is prolonged by the load of friction required to rotate the f-actin (Yasuda *et al.*, 1998). The rotational rate at 2 mM ATP is 0.1 rps, when the length of f-actin is 4 μm (K_m ATP = 0.8 μM ; Yasuda *et al.*, 1998). After the binding of ATP to T β in a very short time (τ_{ATP}), the chemical energy of ATP hydrolysis should be stored in the $\alpha_3\beta_3\gamma$ -actin complex during τ_{step} (about 3 sec), perhaps as an elastic energy. The maximum energy stored during a $2\pi/3$ rotation is approximately 80 pN/nm, a value close to the free energy of ATP hydrolysis. The actin filament bends very little during the load (see photographs in Noji *et al.*, 1997), and thus the energy is stored in $\alpha_3\beta_3\gamma$. When the elastic energy exceeds a certain threshold value, one ATP molecule bound at the P-loop around IV is released.

Any force applied to a subunit is transferred to the next one only through the contact area (Fig. 2 and Table I). Of course, the stress-strain relations should be calculated by the stress tensor, but the structure of $\alpha_3\beta_3\gamma$ is too complicated to apply the tensor. Thus, more simplified scheme of elastic energy transmission was proposed (Oster and Wang, 2000; Cherepanov *et al.*, 1999).

To estimate the elastic energy in $\alpha_3\beta_3\gamma$ under torque, both Poisson ratio (σ) and the Young's modulus (E , the dimension is Pa = N/m²), of the protein are needed. The torsional strain energy is $U = \pi R^4 E \theta^2 / 8L(1 + \sigma)$, where θ is the torsional angle, L is the length of γ , and R is the

radius of γ . The modulus of elasticity (k) of many proteins reported by Gekko's group (Kamiyama and Gekko, 2000), will be useful to estimate E using the equation: $E = 3k(1 - 2\sigma)$. The Poisson ratio of about 0.4 for proteins is acceptable because wide varieties of proteinlike substances, such as rubber (0.46–0.49) and polystyrene (0.34), (Rikantenpyou, Maruzen, Tokyo, 1973) show similar values. The ligands induce a large change in the elasticity of protein as shown by Gekko's group (Kamiyama and Gekko, 2000). Junge's group adopted E of actin (1.8×10^9 N/m²) (Yanagida *et al.*, 2000), but E of a collagen triple helix (5×10^9 N/m²) (Hoffmann *et al.*, 1984) is better, because of the interaction between the α -helices in F_1 (personal communication, T. Yanagida, 2000). Since the torsional strain energy is proportional to E , the estimated value of U is about 2.8-fold of that reported by Junge's group (50 kJ per $2\pi/3$ rotation) (Fig. 1 of Cherepanov *et al.*, 1999).

However, if $\alpha_3\beta_3\gamma$ is subjected to an external torque of 42 pN/nm, the average force acting on the α/β interfaces of about 2 nm from the γ -axis will be 21 pN. The torque of γ is applied through three points of γ (Fig. 3, blue circles, γ 1–3), mainly to β_E at γ_2 and to β_T at γ_3 (Abrahams *et al.*, 1994). The torsional strain applied at γ_2 to VIII and transferred to III–VII, but not to I, I', and II, which are immobile and connected to III via a slender main chain (Fig. 3, red*). The strain applied at γ_3 (the most eccentric portion of $M\gamma$ RGL sequence) to the DELSE(D)ED sequence (Fig. 3) of β_T is propagated from IX–XI to IV, which moves DELSE(D)ED away from γ , and the hinge is open. The elasticity of DELSEED was demonstrated by the fact that any of the residues E β D390, E β D391, and E β S393 in that sequence, can be cross linked to the same residue in the γ subunit (Kagawa, 1999, references therein). Thus, the elastic conformational change at P-loop will release ATP (Oster and Wang, 2000; Kagawa, 1999). Elasticity around the P-loop was shown by cross linking the $\alpha\beta$ interface with a very short diazidonucleotide (Schäfer *et al.*, 1995). The binding of Mg-ATP to $M\beta_E$ at K162, K163, E192, E199, and D256 around the P- and GER-loops bring β_R260 of $M\beta_T$ close to $\alpha F299$ and $\alpha S344$ of $M\alpha$. This conformational signal is transferred through H bonds between Y300, R304, and Y244 of $M\alpha$ (Ren and Allison, 2000). However, correlating these residue interactions with local elastic energy is difficult.

Elastic energy may be stable at least for a few seconds of τ_{step} . The length of τ_{step} can be extended if the proton-driven rotation is stopped by removing ADP + P_i (so-called state 4 respiration). Because of uncoupling (by uncoupling proteins etc.), decoupling, and noncoupling, estimating the stability of elastic energy is difficult in the mitochondrial membrane (Kagawa *et al.*, 1999). Dissoci-

ation of $\alpha_3\beta_3$ into $\alpha_1\beta_1$ by the ATPase reaction has been analyzed using a synchrotron (Sato *et al.*, 1995), and the time course of dissociation was in the same range as the rotation of $\alpha_3\beta_3\gamma$. The elastic energy stored during the domain motions in protein has been discussed in detail (Hayward, 1999).

CONCLUSION

X-ray crystallography of MF_1 and $T\alpha_3\beta_3$, together with domain cross-linking revealed the domain motions that release ATP-Mg from β_T . The α/β interfaces of β constitute the pathway of signal and energy in F_1 (Table I and Fig. 2). ATP synthesis/hydrolysis is dependent on the $\alpha\beta$ interaction at catalytic III to VIII in β_E , and I, I', and II fix the barrel domains for the stator (Fig. 3). Rotation of γ in $\alpha_3\beta_3$ with torque of 42 pN/nm applies pressure on catalytic β_E at γ_2 and at β_T , at γ_3 , etc., cause the hinge motions at H179, G180, and G181, and then at V162 and G163 at IV of the P-loop of $T\beta$ (Masaïke *et al.*, 2000). During the rotation, the contact areas are increased from 7 (93 atoms in $M\beta_E$) to 11 (176 atoms in $M\beta_D$) (catalytic IV, IX, X, and XI, and noncatalytic VIII and IX) (Table I). If the entrance of the P-loop is cross linked, the activity is lost because hinge motion is lost (Ren *et al.*, 1999). The strain is distributed at the α/β interfaces including IV, V, and VIII near the P-loop, GER-loop, and switch domain, respectively, of β . This brings β_R260 of $M\beta_T$ close to $\alpha F299$ and $\alpha S344$ of $M\alpha$ and the conformational signal is transferred through H-bonds in $M\alpha$.

The models of elastic energy in F_0F_1 (Oster and Wang, 2000; Cherepanov *et al.*, 1999) are still contradictory to the experimental results and are not related to the actual domain movements in F_1 . The following experimental results are important.

1. Single-stalk F_0F_1 (Fig. 1 right). X-ray crystallography (Stock *et al.*, 1999) and electron microscopy (Kagawa and Racker, 1966b) of some F_0F_1 preparation show only one stalk. The rotation of F_0 - c observed (Sambongi *et al.*, 1999) may be caused by loss of the stator (Tsunoda *et al.*, 2000). X-ray crystallography indicates that number of the c subunits is 10 (Stock *et al.*, 1999). Thus, F_0 is rotated by the $\pi/5$ step, not the $2\pi/3$, if the stator is intact (double stalk F_0F_1). Then, the stoichiometry cannot be an integer but the transient elastic energy will explain the high efficiency of rotation.

2. The open–close motion is the inherent property of the isolated $T\beta$ (Fig. 1, left). X-ray crystallography, IR, and NMR of isolated β revealed that it is in the open form (O), which is converted into the closed form (C) by adding Mg-ATP. This O–C conversion is independent of

interactions with other subunits. The conversion depends on the pivot residues (H179, G180, and G181) (Fig. 3, right). This motion is transferred through α/β interfaces and the elastic rotates γ .

3. CCO and CCC type of F₁. The CCO type of F₁, shown by X-ray crystallography (Abrahams *et al.*, 1994), has been graphically rotated in the calculation of elastic energy. However, ATPase was not active in the conditions for the CCO-type crystal. On the other hand, the CCC-type F₁ (Bianchet *et al.*, 1998; Weber and Senior, 2000) is incompatible with the results of cross linking β - β (Tsunoda *et al.*, 1999), and T β D311-R333 (Ren *et al.*, 1999). Thus, it should include the elastically deformed C (C').

4. Bi- or trisite catalysis in rotation of $\alpha_3\beta_3\gamma$. The unisite reaction conditions at 1 mM ATP is the base of calculation (Oster and Weber, 2000), but unisite catalysis is observed below nanomolar range, and the $2\pi/3$ step rotation of $\alpha_3\beta_3\gamma$ is observed at higher ATP concentrations: about 150 (bisite) and 490 μ M (trisite).

5. H-bonds in ATP binding. The "binding-zipper" is based on the thermal zippering of H bonds between Mg-ATP and P-loop. There are electrostatic force between the negative charges of phosphate and the surrounding guanidium and ammonium. Tritium exchange could not show zipping of H bonds (Noji *et al.*, 1996). The electrostatic energy between residues is several hundredfold larger (Hall and Pavitt, 1984) and long reaching than that in H bond (0.3 nm, directional) and hydrophobic interaction (very near). Electrostatic force between residues (Table I, red and blue letters) is important in elastic displacement during the rotation.

6. Hardness of the structure. TF₁ is harder than mesophilic F₁s, but this stability is mainly limited to the immobile domains I, I', and II. The relevant residues in III-XI and the hinge accepting the elastic strain are conserved among F₁s.

Finally, these residue movements are not yet expressed by the stress tensor inside $\alpha_3\beta_3\gamma$. X-ray crystallography has been regarded as a purely static technique, and models of elastic energy in F₀F₁ discussed here are necessarily speculative. However, time-resolved crystallography (Moffat, 1998) of F₁ during the photolysis of caged ATP will overcome the limitations of the present X-ray data to analyze the domain motions.

ACKNOWLEDGMENTS

We thank Dr. Y. Shirahara of the National Institute of Genetics for calculating the residue pairs within 0.40 nm across the catalytic and noncatalytic interfaces of the α and β subunits of TF₁ and MFM₁. Thanks are

due to Prof. K. Miki of Kyoto University for providing unpublished data about the T β monomer. We also thank Prof. M. Yoshida at Tokyo Institute of Technology, Prof. M. Futai at Osaka University, Prof. H. Kihara of Kansai Medical University, and Dr. E. Muneyuki at Tokyo Institute of Technology for discussing their studies of TF₁. This work was supported by Grants in Aid for High Technology Research Center from the Ministry of Education, Science and Culture of Japan.

REFERENCES

- Abrahams, J. P., Leslie, A. G. W., Lutter, R. and Walker, J. E. (1994). *Nature(London)* **370**, 621-628.
- Aloise, P., Kagawa, Y., and Coleman, P. S. (1991). *J. Biol. Chem.* **266**, 10368-10376.
- Amano, T., Hisabori, T., Muneyuki, E., and Yoshida, M. (1996). *J. Biol. Chem.* **271**, 18127-18133.
- Amano, T., Matsui, T., Muneyuki, E., Noji, H., Hara, K., Yoshida, M., and Hisabori, T. (1999). *Biochem. J.* **343**, 135-138.
- Arai, M., and Kuwajima, K. (2000). *Advan. Protein Chem.* **53**, 209-282.
- Argos, P., Rossmann, M. G., Grqu, U. M., Zuber, H., Frank, G., and Tratschin, J. D. (1979). *Biochemistry* **18**, 5698-5703.
- Bald, D., Amano, T., Muneyuki, E., Pitard, B., Rigaud, J-L., Kruij, J., Hisabori, T., Yoshida, M., and Shibata, M. (1998). *J. Biol. Chem.* **273**, 865-870.
- Bianchet, M. A., Hullihen, J., Pedersen, P. L., and Amzel, L. M. (1998). *Proc. Natl. Acad. Sci. USA* **95**, 11065-11070.
- Boyer, P. D. (1997) *Annu. Rev. Biochem.* **66**, 717-749.
- Burgard, S., Nett, J. H., Sauer, H. E., Kagawa, Y., Schäfer, H-J., Wise, J. G., Vogel, P. D., and Trommer, W. E. (1994). *J. Biol. Chem.* **269**, 17815-17819.
- Cherepanov, D. A., Mulikidjanian, A. Y., and Junge, W. (1999). *FEBS Lett.* **449**, 1-6.
- Collinson, I. R., Skehel, J. M., Fearnley, I. M., Runswick, J. M., and Walker, J. E. (1996). *Biochemistry* **35**, 12640-12646.
- Futai, M., and Omote, H. (1996). *J. Bioenerg. Biomembr.* **28**, 409-414.
- Futai, M., Noumi, T., and Maeda, M. (1989). *Annu. Rev. Biochem.* **58**, 111-136.
- Garcia, J. J., and Capaldi, R. A. (1998). *J. Biol. Chem.* **273**, 15940-15945.
- Hall, D., and Pravitt, N. (1984). *Compt. Chem.* **5**, 441-450.
- Hayward, S. (1999). *Proteins* **36**, 425-435.
- Hirata, H., Ohno, K., Sone, N., Kagawa, Y., and Hamamoto, T. (1986). *J. Biol. Chem.* **261**, 8939-8943.
- Hofmann, H., Voss, T., Kuhn, K., and Engel, J. (1984). *J. Mol. Biol.* **172**, 325-343.
- Ichida, M., Hakamata, Y., Hayakawa, H., Ueno, E., Ikeda, U., Shimada, K., Hamamoto, T., Kagawa, Y., and Endo, H. (2000). *J. Biol. Chem.* **275**, 15992-16001.
- Jaenicke, R., and Boehm, G. (1998). *Curr. Opinion Struct. Biol.* **8**, 738-748.
- Jault, J. M., Matsui, T., Jault, F. M., Kaibara, C., Muneyuki, E., Yoshida, M., Kagawa, Y., and Allison, W. S. (1995). *Biochemistry* **34**, 16412-16418.
- Jones, P. C., Jiang, W., and Fillingame, R. H. (1998). *J. Biol. Chem.* **273**, 17178-17185.
- Kagawa, Y. (1999). *Advan. Biophys.* **36**, 1-25.
- Kagawa, Y., and Hamamoto, T. (1996). *J. Bioenerg. Biomembr.* **28**, 421-431.
- Kagawa, Y., and Hamamoto, T. (1997). *Biochem. Biophys. Res. Commun.* **240**, 247-256.
- Kagawa, Y., and Racker, E. (1966a). *J. Biol. Chem.* **241**, 2467-2474.

- Kagawa, Y., and Racker, E. (1966b). *J. Biol. Chem.* **241**, 2475–2482.
- Kagawa, Y., and Racker, E. (1971). *J. Biol. Chem.* **246**, 5477–5487.
- Kagawa, Y., Ishizuka, M., Saishu, T., and Nakao, S. (1986). *J. Biochem.* **100**, 923–934.
- Kagawa, Y., Ohta, S., Harada, M., Sato, M., and Ito, Y. (1992). *Ann. N.Y. Acad. Sci.* **671**, 366–376.
- Kagawa, Y., Hamamoto, T., Endo, H., Ichida, M., Shibui, H., and Hayakawa, M. (1997). *Biosci. Rept.* **17**, 115–146.
- Kagawa, Y., Cha, S.-H., Hasegawa, K., Hamamoto, T., and Endo, H. (1999). *Biochem. Biophys. Res. Commun.* **266**, 662–667.
- Kaibara, C., Matsui, T., Hisabori, T., and Yoshida, M. (1996). *J. Biol. Chem.* **271**, 2433–2438.
- Kamiyama, T., and Gekko, K. (2000). *Biochim. Biophys. Acta* **1478**, 257–266.
- Kinoshita, K. Jr. (1999). *FASEB J. Suppl.* **2**, S201–S208.
- Masaie, T., Mitome, N., Noji, H., Muneyuki, E., Yasuda, R., Kinoshita, K., Jr., and Yoshida, M. (2000). *J. Exp. Biol.* **203**, 1–8.
- Le, N. P., Omote, H., Wada, Y., Al-Shawi, M. K., Nakamoto, R. K., and Futai, M. (2000). *Biochemistry* **39**, 2778–2783.
- Leslie, A. G. W., and Walker, J. E. (2000). *Phil. Trans. R. Soc. Lond. B* **355**, 465–472.
- Mitchell, P. (1979) *Science* **206**, 1148–1159.
- Moffat, K. (1998). *Acta Crystallogr. A.* **54**, 833–841.
- Noji, H., Amano, T., and Yoshida, M. (1996). *J. Bioenerg. Biomembr.* **28**, 451–457.
- Noji, H., Yasuda, R., Yoshida, M., and Kinoshita, K. Jr. (1997). *Nature (London)* **386**, 299–302.
- Ohta, S., and Kagawa, Y. (1986). *J. Biochem.* **99**, 135–141.
- Ohta, S., Tsuboi, M., Yoshida, M., and Kagawa, Y. (1980). *Biochemistry* **19**, 2160–2168.
- Omote, H., Nga, P. L., Park, M.-Y., Maeda, M., and Futai, M. (1995). *J. Biol. Chem.* **270**, 25656–25660.
- Omote, H., Tainaka, K., Fujie, K., Iwamoto-Kihara, A., Wada, Y., and Futai, M. (1998). *Arch. Biochem. Biophys.* **358**, 277–282.
- Oster, G., and Wang, H. (2000). *Biochim. Biophys. Acta* **1458**, 482–510.
- Paik, S. R., Yokoyama, K., Yoshida, M., Ohta, T., Kagawa, Y., and Allison, W. S. (1993). *J. Bioenerg. Biomembr.* **25**, 679–684.
- Pedersen, P. L. (1996). *J. Bioenerg. Biomembr.* **28**, 389–395.
- Ren, H., and Allison, W. S. (2000). *J. Biol. Chem.* **275**, 10057–10063.
- Ren, H., Dou, C., Stelzer, M. S., and Allison, W. S. (1999). *J. Biol. Chem.* **274**, 31366–31372.
- Saika, K., and Yoshida, M. (1995). *FEBS Lett.* **368**, 207–210.
- Saika, K., Inaka, K., Matsui, T., Yoshida, M., and Miki, K. (1994). *J. Mol. Biol.* **242**, 709–711.
- Sambongi, Y., Iko, Y., Tanabe, M., Omote, H., Iwamoto-Kihara, A., Ueda, I., Yanagida, T., Wada, Y., and Futai, M. (1999). *Science* **286**, 1687–1688.
- Sasaki, N., Shimada, T., and Sutoh, K. (1998). *J. Biol. Chem.* **273**, 20334–20340.
- Sato, M., Ito, Y., Harada, M., Kihara, H., Tsuruta, H. S., Ohta, S., and Kagawa, Y. (1995). *J. Biochem.* **117**, 113–119.
- Schäfer, H.-J., Rathgeber, G., and Kagawa, Y. (1995). *FEBS Lett.* **377**, 408–412.
- Senter, P., Eckstein, F., and Kagawa, Y. (1983). *Biochemistry* **22**, 5514–5518.
- Shibui, H., Hamamoto, T., Yohda, M., and Kagawa, Y. (1997). *Biochem. Biophys. Res. Commun.* **234**, 341–345.
- Shirakihara, Y., Leslie, A. G. W., Abrahams, J. P., Walker, J. E., Ueda, T., Sekimoto, Y., Kambara, M., Saika, K., Kagawa, Y., and Yoshida, M. (1997). *Structure* **5**, 825–836.
- Stock, D., Leslie, A. G. W., and Walker, J. E. (1999). *Science* **286**, 1700–1705.
- Tozawa, K., Sekino, N., Soga, M., Yagi, H., Yoshida, M., and Akutsu, H. (1995). *FEBS Lett.* **376**, 190–194.
- Tsunoda, S. P., Aggeler, R., Noji, H., Kinoshita, K., Jr., Yoshida, M., and Capaldi, R. A. (2000). *FEBS Lett.* **470**, 244–248.
- Tsunoda, S. P., Muneyuki, E., Amano, T., Yoshida, M., and Noji, H. (1999). *J. Biol. Chem.* **274**, 5701–5706.
- Wang, Z. G., and Ackerman, S. H. (2000). *J. Biol. Chem.* **275**, 5767–5772.
- Weber, J., and Senior, A. E. (2000). *Biochim. Biophys. Acta* **1458**, 300–309.
- Wilkins, S., and Capaldi, R. A. (1998). *Biochim. Biophys. Acta* **1365**, 93–97.
- Xu, T., Zanotti, F., Gaballo, A., Raho, G., and Papa, S. (2000). *Eur. J. Biochem.* **267**, 4445–4455.
- Yagi, H., Tozawa, K., Sekino, N., Iwabuchi, T., Yoshida, M., and Akutsu, H. (1999). *Biophys. J.* **77**, 2175–2183.
- Yan, B., Zhang, W., Ding, J., and Arnold, E. (1999). *J. Protein Chem.* **18**, 807–811.
- Yanagida, T., Kinanuma, K., Tanaka, H., Hikikoshi-Iwane, A., and Esaki, S. (2000). *Curr. Opin. Cell Biol.* **12**, 20–25.
- Yasuda, R., Noji, H., Kinoshita, K. Jr., and Yoshida, M. (1998). *Cell* **93**, 1117–1124.
- Yohda, M., Kagawa, Y., and Yoshida, M. (1986). *Biochim. Biophys. Acta* **850**, 429–435.
- Yohda, M., Ohta, S., Hisabori, T., and Kagawa, Y. (1988). *Biochim. Biophys. Acta* **933**, 156–164.
- Yoshida, M., Sone, N., Hirata, H., and Kagawa, Y. (1975). *J. Biol. Chem.* **250**, 7910–7915.
- Yoshida, M., Sone, N., Hirata, H., Kagawa, Y., and Ui, N. (1979). *J. Biol. Chem.* **254**, 9525–9533.
- Yoshida, T., Yohda, M., Iida, T., Maruyama, T., Taguchi, H., Yazaki, K., Ohta, T., Okada, M., Endo, I., and Kagawa, Y. (1997). *J. Mol. Biol.* **273**, 635–645.